

PHYS 4310, Quiz 2

1. Momentum, \vec{p} , is a vector. Reconcile that notion with the de Broglie relationship which equates momentum to h/λ .

$$\vec{p} = \hbar \vec{k} = \frac{h}{\lambda} \hat{k}$$

h/λ is the magnitude of \vec{p} . The direction of propagation defines the direction of \vec{k} ($\hat{k} = \frac{\vec{k}}{|\vec{k}|}$), which is not explicitly captured in the numerical relation $p = h/\lambda$.

2. In 1773, Benjamin Franklin investigated how oil could "calm" water near a pond in London. He found that a few cm^3 of oil (use 3 cm^3) could cover an area of 2000 m^2 . Assuming that the oil on the surface of the water is one molecule thick, estimate the size of an oil molecule. As a historical aside, Franklin never did this calculation and thus never obtained an estimate for the linear dimension of a single molecule. However, similar methods were employed nearly a century later by Lord Kelvin and other scientists to obtain molecule size estimates (French and Taylor, 1978).

$$\begin{aligned} V &= 3 \text{ cm}^3 \\ &= 3 \text{ cm}^3 \times \left(\frac{10^{-2} \text{ m}}{1 \text{ cm}} \right)^3 \\ &= 3 \times 10^{-6} \text{ m}^3 \end{aligned}$$

$$\begin{aligned} V &= A \cdot d = 3 \times 10^{-6} \text{ m}^3 \\ A &= 2 \times 10^3 \text{ m}^2 \end{aligned}$$

$$d = \frac{3}{2} 10^{-9} \text{ m} = 1.5 \text{ nm}$$

This estimate is quite low (especially for long chain oil molecules), but not unreasonable.