

PHYS 4310, Homework 9 due on Monday, November 18th, 2015 at 5 pm

Griffiths (2nd edition): 4.18 (15 points)

2. (20 points) Using the Schrödinger equation for $l = 0$ and $V(r) = -\frac{Ze^2}{r}$ (cgs units) with $u = r\psi$ (for a hydrogen atom), substitute in an assumed form of $u(r)$: $u(r) = (Ar + Br^2)e^{-br}$. (*Hint*: use equation 4.53 for $l = 0$ and replace the V term with the one specified here.)

- Find the values of b and the ratio of B/A using this form of $u(r)$.
- Verify that it corresponds to the second energy level of hydrogen with E equal to $Z^2/4$ times the ground state of hydrogen and with $B/A = -Z/2a_0$ (where a_0 is the Bohr radius).
- What is the value of the coefficient b in terms of a_0 ?

3. (25 points) An important process in nuclear reactors and in nuclear weapons is the decay of tritium (${}^3\text{H}$ or H with two neutrons) into ionized ${}^3\text{He}$ via beta decay. This decay releases 18.6 keV per reaction with the ionized He going through another nuclear reaction (this time with a neutron) to produce ${}^1\text{H}$ and ${}^3\text{H}$ (or hydrogen and tritium). Consider the probability, P , that tritium will decay into ionized He by approximating the tritium wavefunction as the ground state of hydrogen and the ionized He wavefunction as the ground state of hydrogen with $Z = 2$.

- What is $P = |\langle\psi_{\text{He}}|\psi_{\text{T}}\rangle|^2$? I'm looking for both the explicit form of the integral and the computationally calculated value. *Hint*: P is between 0.5 and 1.0.
- What is P if the ψ_{He} is in $l=1, m=0$ (excited state)?

4. (20 points) Consider a particle of mass m in a central potential:

$$V(r) = \frac{A}{r^2} + \frac{B}{r}, \quad (1)$$

where $A, B > 0$. Show that the bound states of angular momentum, l , exist with energies:

$$E_{n,l} = -\frac{mB^2}{2\hbar^2(n + \lambda + 1)^2}, \quad (2)$$

for $n = 0, 1, 2, \dots$ with $\lambda(\lambda + 1) = l(l + 1) + 2mA/\hbar^2$. Compare the degeneracy of this spectrum with that of the hydrogen atom ($A = 0, B = e^2$).