

PHYS 4310, Homework 9 due on Monday, November 18<sup>th</sup>, 2015 at 5 pm

Griffiths (2<sup>nd</sup> edition): 4.18 (15 points)

2. (20 points) Using the Schrödinger equation for  $l = 0$  and  $V(r) = -\frac{Ze^2}{r}$  (cgs units) with  $u = r\psi$  (for a hydrogen atom), substitute in an assumed form of  $u(r)$ :  $u(r) = (Ar + Br^2)e^{-br}$ . (*Hint*: use equation 4.53 for  $l = 0$  and replace the  $V$  term with the one specified here.)

- Find the values of  $b$  and the ratio of  $B/A$  using this form of  $u(r)$ .
- Verify that it corresponds to the second energy level of hydrogen with  $E$  equal to  $Z^2/4$  times the ground state of hydrogen and with  $B/A = -Z/2a_0$  (where  $a_0$  is the Bohr radius).
- What is the value of the coefficient  $b$  in terms of  $a_0$ ?

3. (25 points) An important process in nuclear reactors and in nuclear weapons is the decay of tritium ( $^3\text{H}$  or  $\text{H}$  with two neutrons) into ionized  $^3\text{He}$  via beta decay. This decay releases 18.6 keV per reaction with the ionized He going through another nuclear reaction (this time with a neutron) to produce  $^1\text{H}$  and  $^3\text{H}$  (or hydrogen and tritium). Consider the probability,  $P$ , that tritium will decay into ionized He by approximating the tritium wavefunction as the ground state of hydrogen and the ionized He wavefunction as the ground state of hydrogen with  $Z = 2$ .

- What is  $P = |\langle\psi_{\text{He}}|\psi_{\text{T}}\rangle|^2$ ? I'm looking for both the explicit form of the integral and the computationally calculated value. *Hint*:  $P$  is between 0.5 and 1.0.
- What is  $P$  if the  $\psi_{\text{He}}$  is in  $l=1, m=0$  (excited state)?

4. (20 points) Consider a particle of mass  $m$  in a central potential:

$$V(r) = \frac{A}{r^2} + \frac{B}{r}, \quad (1)$$

where  $A, B > 0$ . Show that the bound states of angular momentum,  $l$ , exist with energies:

$$E_{n,l} = -\frac{mB^2}{2\hbar^2(n + \lambda + 1)^2}, \quad (2)$$

for  $n = 0, 1, 2, \dots$  with  $\lambda(\lambda + 1) = l(l + 1) + 2mA/\hbar^2$ . Compare the degeneracy of this spectrum with that of the hydrogen atom ( $A = 0, B = e^2$ ).